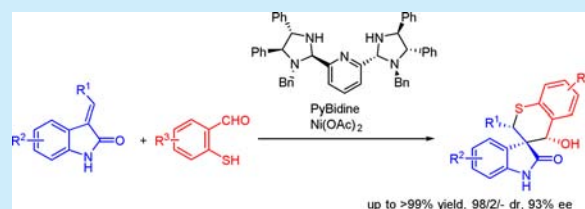


PyBidine–Ni(OAc)<sub>2</sub>-Catalyzed Michael/Aldol Reaction of Methyleneindolinones and Thiosalicylaldehydes for Stereochemically Divergent Thiochromanyl-spirooxindolesTakayoshi Arai,<sup>\*,†</sup> Tomoya Miyazaki,<sup>†</sup> Hiroki Ogawa,<sup>†</sup> and Hyuma Masu<sup>‡</sup><sup>†</sup>Molecular Chirality Research Center, and Department of Chemistry, Graduate School of Science and <sup>‡</sup>Center for Analytical Instrumentation, Chiba University, 1-33 Yayoi, Inage, Chiba 263-8522, Japan

## S Supporting Information

**ABSTRACT:** (*S,S*)-Diphenylethylenediamine-derived bis-(imidazolidine)pyridine (PyBidine)–Ni(OAc)<sub>2</sub> complex catalyzed the asymmetric Michael/aldol reaction of methyleneindolinone and thiosalicylaldehyde to produce (2′*R*,3*S*,4′*R*)-thiochromanyl-spirooxindole having three contiguous stereogenic centers.



Spirooxindoles are complex skeletal structures with significant biological activity that are found in many naturally occurring alkaloids. Spirotryprostatin A isolated from *Aspergillus fumigatus* inhibits the cell cycle at the G2/M phase in a mouse cdc2 mutant cell line, tsFT210 (Figure 1).<sup>1</sup> Gelsemium

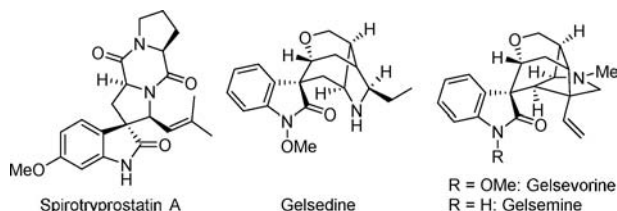


Figure 1. Examples of biologically significant spirooxindoles.

spirooxindole alkaloids (e.g., glesedine, gelsevorine, and gelsemine) isolated from the ancient medicine yakatsu stored in the Shosoin Repository exhibit a wide range of biological activities, including analgesic, antiinflammatory, and antitumor effects.<sup>2</sup>

Among a wide range of approaches directed toward the spirooxindole structure, synthesis using catalytic asymmetric reaction can provide a powerful methodology for accessing novel pharmacophores, not just the naturally occurring compounds.<sup>3,4</sup> For example, artificial pyrrolidinyl-spirooxindoles, which contain a five-membered ring, have been constructed using the catalytic asymmetric [3 + 2] cycloaddition of methyleneindolinones and imino esters.<sup>5</sup> For catalytic access to spirooxindole structures with a six-membered heterocycle, asymmetric tandem Michael/Michael reactions using methyleneindolinones have been investigated, and Zhao has pioneered a thiourea-catalyzed synthesis of tetrahydroquinolyl-spirooxindoles.<sup>6a</sup>

Zhu also reported the asymmetric syntheses of tetrahydroquinolyl-spirooxindoles using a (DHQD)<sub>2</sub>PHAL catalyst

and chromanyl-3,3′-oxindoles using a squaramide catalyst.<sup>7</sup> Quite recently, Enders succeeded in the catalytic asymmetric synthesis of chromanyloxindoles by a chiral thiourea-catalyzed oxa-Michael/1,6-addition.<sup>8,9</sup> However, successful syntheses of sulfur-containing chiral versions are limited, although replacement of the oxygen atom by sulfur can be a rational strategy for improving pharmacological activity.<sup>6b,10,11</sup>

The present report describes the first catalytic asymmetric synthesis of thiochromanyl-spirooxindoles.<sup>12</sup> A program on diversity-oriented asymmetric catalysis (DOAC) led to the catalytic asymmetric synthesis of stereochemically diversified pyrrolidinyl-spirooxindoles. An imidazoline aminophenol (IAP)–Ni(OAc)<sub>2</sub> promoted the first general *exo*′-selective [3 + 2] cycloaddition (Scheme 1 (1)).<sup>13a</sup> Bis(imidazolidine)pyridine (PyBidine)–Cu(OTf)<sub>2</sub> was used as the catalyst for the highly *endo*-selective [3 + 2] cycloaddition of methyleneindolinone and iminoesters (Scheme 1 (2)).<sup>13b</sup> The DOAC of thiochromanes also was achieved using the IAP–Ni(OAc)<sub>2</sub> catalyst (Scheme 1 (3)).<sup>14</sup>

By combining the chemistry shown in Scheme 1, the synthesis of chiral thiochromanyl-spirooxindoles was investigated by determining the appropriate catalyst for the reaction of asymmetric Michael/aldol reaction of methyleneindolinone (1a) and thiosalicylaldehyde (2a) (Table 1).

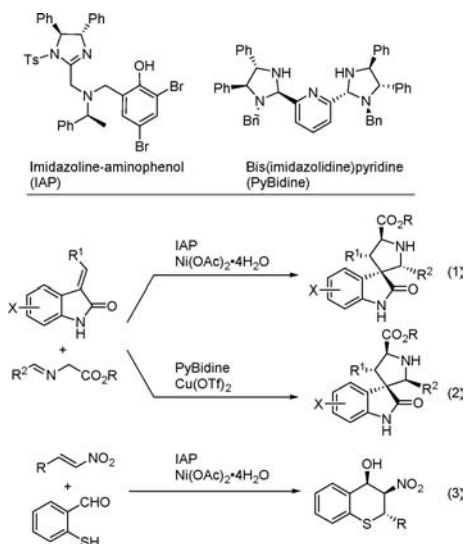
When the IAP–Ni(OAc)<sub>2</sub> catalyst was used in the reaction of 1a and 2a, the reaction proceeded at −78 °C to give the desired thiochromanyl-spirooxindole (3a) in a diastereoselective manner; however, the major diastereomer was obtained in racemic form (Table 1, entry 1). Although an effort was made to obtain optically active 3a, promoting the asymmetric induction for 3a using the IAP–metal catalyst was very difficult. In contrast, the use of PyBidine–Cu(OTf)<sub>2</sub>, which was effective in the *endo*-selective [3 + 2] cycloaddition of

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**Scheme 1. IAP- and PyBidine-Metal-Catalyzed Asymmetric Synthesis of Pyrrolidinyl-spirooxindoles and Thiochromanes**



**Table 1. Catalyst Exploration for Michael/Aldol Reaction of Methyleneindolinone and Thiosalicylaldehyde**

entry	ligand	metal salt	yield <sup>a</sup> (%)	diastereomeric ratio 3a/3a'/3a''	ee <sup>b</sup> (%)
1	IAP	Ni(OAc) <sub>2</sub> <sup>c</sup>	48	74/15/11	rac
2	PyBidine	Cu(OTf) <sub>2</sub>	27	64/23/13	−12
3	PyBidine	Cu(OAc) <sub>2</sub>	70	67/15/18	−7
4	PyBidine	Co(OAc) <sub>2</sub> <sup>c</sup>	65	74/7/19	−8
5	PyBidine	Ni(OAc) <sub>2</sub> <sup>c</sup>	79	80/6/14	50
6	PyBidine	Zn(OAc) <sub>2</sub>	99	74/10/16	4
7 <sup>d</sup>	PyBidine	Ni(OAc) <sub>2</sub> <sup>c</sup>	99	82/4/14	66
8 <sup>d,e</sup>	PyBidine	Ni(OAc) <sub>2</sub> <sup>c</sup>	99	93/2/5	88
9 <sup>d,f</sup>	PyBidine	Ni(OAc) <sub>2</sub> <sup>c</sup>	99	91/3/6	76

<sup>a</sup>Combined yield of diastereomers. <sup>b</sup>ee of major diastereomer **3a**. <sup>c</sup>Tetrahydrate was used. <sup>d</sup>Without NEt<sub>3</sub>. <sup>e</sup>Reaction was performed at −40 °C. <sup>f</sup>**2a** was slowly added over 0.5 h.

methyleneindolinones, resulted in the production of **3a** with 12% ee. The PyBidine–Cu(OAc)<sub>2</sub> showed greater catalytic activity and provided the product in 70% yield. Among the acetates of first-row transition metals examined, the PyBidine–Ni(OAc)<sub>2</sub> catalyst gave thiochromanyl-spirooxindoles in a 79% combined yield with a diastereomeric ratio (dr) of 80/6/14 and **3a** as the major isomer with 50% ee (Table 1, entry 5). After optimization (reaction at −40 °C without NEt<sub>3</sub>), the thiochromanyl-spirooxindoles were produced in 99% yield with up to 93/2/5 diastereoselectivity, with **3a** being produced in 88% ee (entry 8). (Detailed optimization conditions are available in the Supporting Information.)

The applicability of the PyBidine–Ni(OAc)<sub>2</sub>-catalyzed asymmetric synthesis of thiochromanyl-spirooxindoles is outlined in Table 2.

**Table 2. Asymmetric Michael/Aldol Reaction of Methyleneindolinones and Thiosalicylaldehydes Using PyBidine–Ni(OAc)<sub>2</sub> Catalyst**

entry	ligand	metal salt	yield <sup>a</sup> (%)	diastereomeric ratio 3a/3a'/3a''	ee <sup>b</sup> (%)
3a	PyBidine	Ni(OAc) <sub>2</sub> ·4H <sub>2</sub> O	99%	93/2/5	88% ee
3b	PyBidine	Ni(OAc) <sub>2</sub> ·4H <sub>2</sub> O	94%	93/2/5	69% ee
3c	PyBidine	Ni(OAc) <sub>2</sub> ·4H <sub>2</sub> O	>99%	93/3/4	82% ee
3d	PyBidine	Ni(OAc) <sub>2</sub> ·4H <sub>2</sub> O	>99%	86/6/8	69% ee
3e	PyBidine	Ni(OAc) <sub>2</sub> ·4H <sub>2</sub> O	89%	80/16/4	81% ee
3f	PyBidine	Ni(OAc) <sub>2</sub> ·4H <sub>2</sub> O	85%	98/-/2	84% ee
3g	PyBidine	Ni(OAc) <sub>2</sub> ·4H <sub>2</sub> O	98%	74/4/22	84% ee
3h	PyBidine	Ni(OAc) <sub>2</sub> ·4H <sub>2</sub> O	>99%	98/2/-	93% ee
3i	PyBidine	Ni(OAc) <sub>2</sub> ·4H <sub>2</sub> O	>99%	92/8/-	89% ee
3j	PyBidine	Ni(OAc) <sub>2</sub> ·4H <sub>2</sub> O	>99%	93/5/2	54% ee
3k	PyBidine	Ni(OAc) <sub>2</sub> ·4H <sub>2</sub> O	>99%	97/2/1	94% ee
3l	PyBidine	Ni(OAc) <sub>2</sub> ·4H <sub>2</sub> O	>99%	93/3/4	89% ee
3m	PyBidine	Ni(OAc) <sub>2</sub> ·4H <sub>2</sub> O	>99%	87/7/6	87% ee
3n	PyBidine	Ni(OAc) <sub>2</sub> ·4H <sub>2</sub> O	>99%	95/2/3	82% ee
3o	PyBidine	Ni(OAc) <sub>2</sub> ·4H <sub>2</sub> O	73%	98/1/1	90% ee
3p	PyBidine	Ni(OAc) <sub>2</sub> ·4H <sub>2</sub> O	>99%	79/9/12	9% ee
3q	PyBidine	Ni(OAc) <sub>2</sub> ·4H <sub>2</sub> O	92%	65/17/18	5% ee

<sup>a</sup>**2a** was slowly added over 0.5 h.

For aryl-substituted methyleneindolinones, both electron-deficient and -donating groups on the benzene ring were used successfully to yield thiochromanyl-spirooxindoles with high diastereoselectivity, and **3h** with an *o*-Br substituent was obtained with up to 93% ee. Slow addition of **2a** over 0.5 h gave better asymmetric induction when using electron-deficient methyleneindolinones to give **3c** and **3d**. Alkyl-substituted methyleneindolinone also was compatible and yielded **3i** with 89% ee. Introduction of chloro and methyl groups on the indolinone ring provided **3k–m**. Furthermore, substituted



thiosalicylaldehydes gave **3n** and **3o** with 82% ee and 90% ee, respectively.

The absolute structure of the main diastereomer was determined by single-crystal X-ray crystallography of **3h** (Figure 2). Using the (*S,S*)-diphenylethylenediamine-derived PyBi-

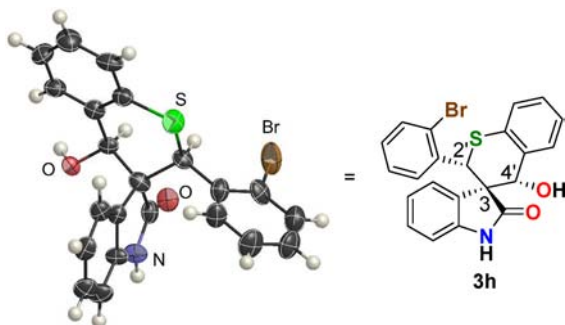
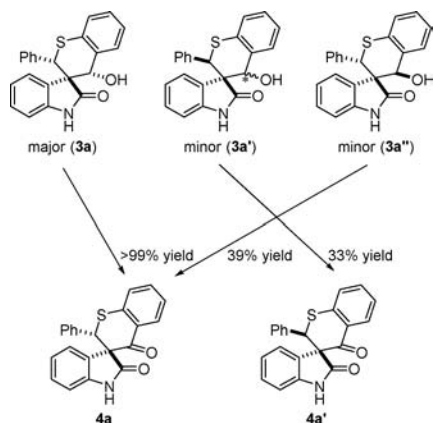


Figure 2. X-ray structure of major diastereomer **3h**.

dine–Ni(OAc)<sub>2</sub> catalyst, (2'*S*,3*S*,4'*R*)-4'-hydroxy-2'-phenylspiro[indoline-3,3'-thiochroman]-2-one (**3h**) was obtained as the major product.

To obtain stereochemically divergent products from thiochromanyl-spirooxindoles, **3a** and its diastereomers (**3a'** and **3a''**) were oxidized by the Dess–Martin reagent as shown in Scheme 2.

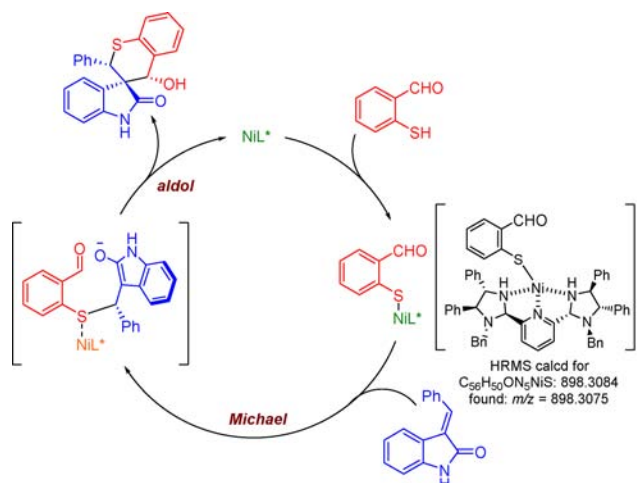
Scheme 2. Oxidation of Michael/Aldol Adducts of Methyleneindolinone (**1a**) and Thiosalicylaldehyde (**2a**) (Dess–Martin Reagent, CH<sub>2</sub>Cl<sub>2</sub>, rt, 17 h)



The oxidation of both **3a** and **3a''** gave the same ketone **4a**, which suggested that **3a''** was an epimer at the 4'-hydroxyl group of **3a**. As expected, oxidation of the remaining isomer **3a'**, which is an epimer at the phenyl position, gave **4a'**.

The proposed catalytic cycle for the PyBidine–Ni(OAc)<sub>2</sub>-catalyzed asymmetric reaction of methyleneindolinone and thiosalicylaldehyde is provided in Scheme 3. When PyBidine–Ni(OAc)<sub>2</sub> was mixed with thiosalicylaldehyde (**2a**), HRMS analysis provided a peak at *m/z* = 898.3075, which corresponded to [PyBidine–Ni–S(C<sub>6</sub>H<sub>4</sub>)CHO]<sup>+</sup>. Due to the strong affinity of the soft thiol functional group for the late transition metal, reaction began with formation of the nickel thiolate of thiosalicylaldehyde. The nickel–thiolate then attacked the methyleneindolinone in a Michael reaction to provide the enolate of indolinone. Further intramolecular aldol

Scheme 3. Proposed Catalytic Cycle



reaction of the nickel enolate furnished the thiochromanyl-spirooxindole with regeneration of the PyBidine–Ni catalyst.

Details of the reaction using the (*S,S*)-diphenylethylenediamine-derived PyBidine–Ni(OAc)<sub>2</sub> catalyst is shown in Figure 3. By following the catalytic cycle, the PyBidine–Ni(OAc)<sub>2</sub>

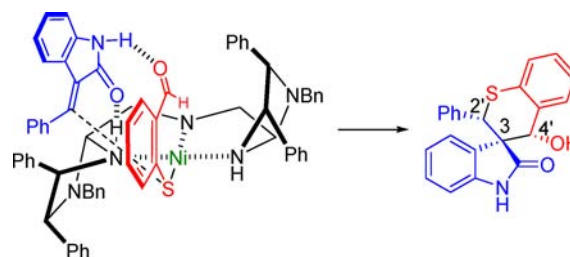


Figure 3. Plausible reaction mode for the yield of major diastereomer **3a**.

complex generated the nickel thiolate. The methyleneindolinone also was activated in the second quadrant of the PyBidine–Ni(OAc)<sub>2</sub> complex using hydrogen bonding through the NH of the imidazolidine ring.<sup>13b</sup> The hydrogen bonding controlled the direction of methyleneindolinone for acceptance of nucleophilic addition of thiolate and guided the following aldol reaction to provide (2'*S*,3*S*,4'*R*)-4'-hydroxy-2'-phenylspiro[indoline-3,3'-thiochroman]-2-one **3** as the major product (Figure 3).

In this plausible reaction model, an additional hydrogen-bonding network between the NH of methyleneindolinone and thiosalicylaldehyde controlled attachment of the 4'*R*-hydroxyl group, which is supported by the low asymmetric induction for giving *N*-acetylated **3p** and *N*-methylated **3q** (Table 2).

In conclusion, this is the first example of a highly diastereoselective asymmetric Michael/aldol reaction of methyleneindolinones and thiosalicylaldehydes catalyzed by bis-(imidazolidine)pyridine (PyBidine)–Ni(OAc)<sub>2</sub> to produce thiochromanyl-spirooxindoles. This provides a wide range of diverse thiochromanyl-spirooxindoles that show promise for application as biologically active pharmaceuticals.



## ■ ASSOCIATED CONTENT

## ■ Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.orglett.6b02783.

Experimental procedure and analytical data (PDF)

X-ray crystallographic data for 3h (CIF)

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## Notes

The authors declare no competing financial interest.

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## ■ REFERENCES

- (1) Cui, C.-B.; Kakeya, H.; Osada, H. *J. Antibiot.* **1996**, *49*, 832.
- (2) Kitajima, M.; Nakamura, T.; Kogure, N.; Ogawa, M.; Mitsuno, Y.; Ono, K.; Yano, S.; Aimi, N.; Takayama, H. *J. Nat. Prod.* **2006**, *69*, 715.
- (3) (a) Franz, A. K.; Hanhan, N. V.; Ball-Jones, N. R. *ACS Catal.* **2013**, *3*, 540. (b) Narayan, R.; Potowski, M.; Jia, Z.-J.; Antonchick, A. P.; Waldmann, H. *Acc. Chem. Res.* **2014**, *47*, 1296. (c) Yu, B.; Yu, D.-Q.; Liu, H.-M.; Xing, H. *Beilstein J. Org. Chem.* **2016**, *12*, 1000.
- (4) For pioneering works, see: (a) Chen, X.-H.; Wei, Q.; Luo, S.-W.; Xiao, H.-W.; Gong, L.-Z. *J. Am. Chem. Soc.* **2009**, *131*, 13819. (b) Antonchick, A. P.; Gerding-Reimers, C.; Catarinella, M.; Schürmann, M.; Preut, H.; Ziegler, S.; Rauh, D.; Waldmann, H. *Nat. Chem.* **2010**, *2*, 735. (c) Liu, T.-L.; Xue, Z.-Y.; Tao, H.-Y.; Wang, C.-J. *Org. Biomol. Chem.* **2011**, *9*, 1980. For recent selected advances, see: (d) Chen, K.-Q.; Li, Y.; Zhang, C.-L.; Sun, D.-Q.; Ye, S. *Org. Biomol. Chem.* **2016**, *14*, 2007. (e) Zhao, K.; Zhi, Y.; Li, X.; Puttreddy, R.; Rissanen, K.; Enders, D. *Chem. Commun.* **2016**, *52*, 2249. (f) Han, X.; Chan, W.-L.; Yao, W.; Wang, Y.; Lu, Y. *Angew. Chem., Int. Ed.* **2016**, *55*, 6492. (g) Sankar, M. G.; Garcia-Castro, M.; Golz, C.; Strohmman, C.; Kumar, K. *Angew. Chem., Int. Ed.* **2016**, *55*, 9709. (h) Zhang, J.-X.; Wang, H.-Y.; Jin, Q.-W.; Zheng, C.-W.; Zhao, G.; Shang, Y.-J. *Org. Lett.* **2016**, *18*, 4774.
- (5) Reviews: (a) Cheng, D.; Ishihara, Y.; Tan, B.; Barbas, C. F., III. *ACS Catal.* **2014**, *4*, 743. (b) Ball-Jones, N. R.; Badillo, J. J.; Franz, A. K. *Org. Biomol. Chem.* **2012**, *10*, 5165. (c) Santos, M. M. *Tetrahedron* **2014**, *70*, 9735.
- (6) (a) Huang, Y.; Zheng, C.; Zhao, G. *RSC Adv.* **2013**, *3*, 16999. (b) Huang, Y.; Zheng, C.; Chai, Z.; Zhao, G. *Adv. Synth. Catal.* **2014**, *356*, 579.
- (7) Mao, H.; Lin, A.; Tang, Y.; Shi, Y.; Hu, H.; Cheng, Y.; Zhu, C. *Org. Lett.* **2013**, *15*, 4062.
- (8) Zhao, K.; Zhi, Y.; Shu, T.; Valkonen, A.; Rissanen, K.; Enders, D. *Angew. Chem., Int. Ed.* **2016**, *55*, 12104.
- (9) For selected examples of catalytic asymmetric construction of a spiro[4,5] system, see the following pioneering works: (a) Bencivenni, G.; Wu, L.; Mazzanti, A.; Giannichi, B.; Pescioli, F.; Song, M.; Bartoli, G.; Melchiorre, P. *Angew. Chem., Int. Ed.* **2009**, *48*, 7200. (b) Jiang, K.; Jia, Z.; Yin, X.; Wu, L.; Chen, Y. *Org. Lett.* **2010**, *12*, 2766. (c) Wei, Q.; Gong, L. *Org. Lett.* **2010**, *12*, 1008. (d) Chen, W.; Wu, Z.; Pei, Q.; Cun, L.; Zhang, X.; Yuan, W. *Org. Lett.* **2010**, *12*, 3132. (e) Companyó, X.; Zea, A.; Alba, A.-N. R.; Mazzanti, A.; Moyano, A.; Rios, R. *Chem. Commun.* **2010**, *46*, 6953. (f) Shen, L.; Shao, P.; Ye, S. *Adv. Synth. Catal.* **2011**, *353*, 1943. (g) Jia, Z.; Jiang, H.; Li, J.; Gschwend, B.; Li, Q.; Yin, X.; Grouleff, J.; Chen, Y.; Jørgensen, K. A. *J. Am. Chem. Soc.* **2011**, *133*, 5053. (h) Cai, Q.; Zheng, C.; Zhang, J.-W.; You, S.-L. *Angew. Chem., Int. Ed.* **2011**, *50*, 8665. (i) Tan, B.; Hernández-Torres, G.; Barbas, C. F., III. *J. Am. Chem. Soc.* **2011**, *133*, 12354. (j) Zhang, X.-N.; Chen, G.-Q.; Dong, X.; Wei, Y.; Shi, M. *Adv. Synth. Catal.* **2013**, *355*, 3351. (k) Li, G.; Liang, T.; Wojtas, L.; Antilla, J. C. *Angew. Chem., Int. Ed.* **2013**, *52*, 4628. (l) Cui, H.-L.; Tanaka, F. *Chem. - Eur. J.* **2013**, *19*, 6213. (m) Yang, W.; Du, D.-M. *Chem. Commun.* **2013**, *49*, 8842. For recent advances, see: (n) Gao, T.-P.; Lin, J.-P.; Hu, X.-Q.; Xu, P.-F. *Chem. Commun.* **2014**, *50*, 8934. (o) Cao, W.; Liu, X.; Guo, J.; Lin, L.; Feng, X. *Chem. - Eur. J.* **2015**, *21*, 1632. (p) Abbaraju, S.; Ramireddy, N.; Rana, N. K.; Arman, H.; Zhao, J. C.-G. *Adv. Synth. Catal.* **2015**, *357*, 2633. (q) Huang, X.; Pham, K.; Yi, W.; Zhang, X.; Clamens, C.; Hyatt, J. H.; Jasinski, J. P.; Tayviah, U.; Zhang, W. *Adv. Synth. Catal.* **2015**, *357*, 3820. (r) Zhang, D.-Y.; Xu, L.; Wu, H.; Gong, L.-Z. *Chem. - Eur. J.* **2015**, *21*, 10314. (s) Zhou, R.; Wu, Q.; Guo, M.; Huang, W.; He, X.; Yang, L.; Peng, F.; He, G.; Han, B. *Chem. Commun.* **2015**, *51*, 13113. (t) Wang, S.; Jiang, Y.; Wu, S.; Dong, G.; Miao, Z.; Zhang, W.; Sheng, C. *Org. Lett.* **2016**, *18*, 1028. (u) Kumarswamyreddy, N.; Kesavan, V. *Org. Lett.* **2016**, *18*, 1354. (v) Han, J.-J.; Chang, C.-H. *Chem. Commun.* **2016**, *52*, 2322. (w) Wu, J.-L.; Du, B.-X.; Zhang, Y.-C.; He, Y.-Y.; Wang, J.-Y.; Wu, P.; Shi, F. *Adv. Synth. Catal.* **2016**, *358*, 2777.
- (10) Recent design of sulfur-containing molecules: (a) Kanbe, Y.; Kim, M.-H.; Nishimoto, M.; Ohtake, Y.; Tsunenari, T.; Taniguchi, K.; Ohizumi, I.; Kaiho, S.; Nabuchi, Y.; Kawata, Y.; Morikawa, K.; Jo, J.-C.; Kwon, H.-A.; Lim, H.-S.; Kim, H.-Y. *Bioorg. Med. Chem. Lett.* **2006**, *16*, 4090. (b) Jia, W.; Liu, Y.; Li, W.; Liu, Y.; Zhang, D.; Zhang, P.; Gong, P. *Bioorg. Med. Chem.* **2009**, *17*, 4569. (c) Yoneya, T.; Taniguchi, K.; Nakamura, R.; Tsunenari, T.; Ohizumi, I.; Kanbe, Y.; Morikawa, K.; Kaiho, S.-I.; Yamada-Okabe, H. *Anticancer Res.* **2010**, *30*, 873. (d) Xu, S.-Q.; Yan, X.; Chen, Y.; Xia, P.; Qian, K.; Yu, D.; Xia, Y.; Yang, Z.-Y.; Morris-Natschke, S. L.; Lee, K.-H. *Bioorg. Med. Chem.* **2010**, *18*, 7203. (e) Liu, H.-S.; Xu, S.-Q.; Cheng, M.; Chen, Y.; Xia, P.; Qian, K.; Xia, Y.; Yang, Z.-Y.; Chen, C.-H.; Morris-Natschke, S. L.; Lee, K.-H. *Bioorg. Med. Chem. Lett.* **2011**, *21*, 5831.
- (11) Selected examples for catalytic asymmetric construction of thiochromane and thiochromene skeletons, see: (a) Wang, W.; Li, H.; Wang, J.; Zu, L. *J. Am. Chem. Soc.* **2006**, *128*, 10354. (b) Zu, L.; Wang, J.; Li, H.; Xie, H.; Jiang, W.; Wang, W. *J. Am. Chem. Soc.* **2007**, *129*, 1036. (c) Zu, L.; Xie, H.; Li, H.; Wang, J.; Jiang, W.; Wang, W. *Adv. Synth. Catal.* **2007**, *349*, 1882. (d) Zhao, G. - L.; Vesely, J.; Rios, R.; Ibrahim, I.; Sünden, H.; Córdova, A. *Adv. Synth. Catal.* **2008**, *350*, 237. (e) Dodda, R.; Goldman, J. J.; Mandal, T.; Zhao, C. - G.; Broker, G. A.; Tiekink, E. R. T. *Adv. Synth. Catal.* **2008**, *350*, 537. (f) Wang, J.; Xie, H.; Li, H.; Zu, L.; Wang, W. *Angew. Chem., Int. Ed.* **2008**, *47*, 4177. (g) Gao, Y.; Ren, Q.; Wu, H.; Li, M.; Wang, J. *Chem. Commun.* **2010**, *46*, 9232. (h) Liu, T.-L.; He, Z.-L.; Wang, C.-J. *Chem. Commun.* **2011**, *47*, 9600. (i) Dong, X.-Q.; Fang, X.; Tao, H.-Y.; Zhou, X.; Wang, C.-J. *Chem. Commun.* **2012**, *48*, 7238. (j) Choudhury, A. R.; Mukherjee, S. *Adv. Synth. Catal.* **2013**, *355*, 1989. (k) Zhao, B.-L.; Du, D.-M. *Asian J. Org. Chem.* **2015**, *4*, 778. (l) Ahlemeyer, N. A.; Birman, V. B. *Org. Lett.* **2016**, *18*, 3454.
- (12) (a) Duan, S.-W.; Li, Y.; Liu, Y.-Y.; Zou, Y.-Q.; Shi, D.-Q.; Xiao, W.-J. *Chem. Commun.* **2012**, *48*, 5160. (b) Wang, S.; Jiang, Y.; Wu, S.; Dong, G.; Miao, Z.; Zhang, W.; Sheng, C. *Org. Lett.* **2016**, *18*, 1028.
- (13) (a) Awata, A.; Arai, T. *Chem. - Eur. J.* **2012**, *18*, 8278. (b) Arai, T.; Ogawa, H.; Awata, A.; Sato, M.; Watabe, M.; Yamanaka, M. *Angew. Chem., Int. Ed.* **2015**, *54*, 1595.
- (14) Arai, T.; Yamamoto, Y. *Org. Lett.* **2014**, *16*, 1700.